

# CLASSIFICATION & NOMENCLATURE

## Format for IUPAC name :

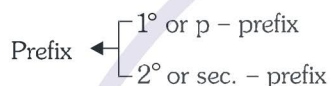
<u>s - prefix</u>	+	<u>p - prefix</u>	+	<u>word root</u>	+	<u>p - suffix</u>	+	<u>s - suffix</u>
Substituents with locants		cyclo		Alk word according to carbon in parent C chain		- ane - ene - yne		According to main functional group given in priority table

(a) **Locant** : Locants are link by (,) comma.

- Locants and alphabets are separated by hyphen (-). [2, 3 - dimethyl pentane]
- di, tri, iso, neo and cyclo are neither separated by comma nor by hyphen

(b) **Prefix** :- According to substituents .

Prefix (es) are written in alphabetical order before root word.



Cyclo is  $1^\circ$  prefix and used for cyclic compound.

$2^\circ$  prefix is used for substituents and written before  $1^\circ$  prefix.

**For acyclic compounds :**  $2^\circ$  prefix + Root word +  $1^\circ$  suffix +  $2^\circ$  suffix.

Substituents	Prefix	Substituents	Prefix
- R	Alkyl group	- OR	Alkoxy
- X (F, Cl, Br, I)	Halo	- $\text{N} \begin{array}{l} \text{=O} \\ \text{O} \end{array}$	Nitro
- O - N=O	Nitrite	- N = O	Nitroso
- CH <sub>2</sub> OH	Hydroxymethyl	- CH <sub>2</sub> Cl	Chloromethyl
- NHC <sub>2</sub> H <sub>5</sub>	Ethylamino		

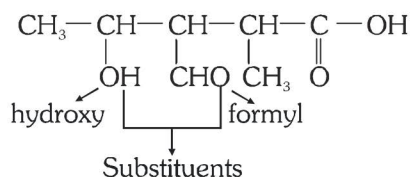
(c) **Word root** : According to number of carbons in parent C-chain.

Number of carbons	Root word	Number of carbons	Root word	Number of carbons	Root word
1	Meth	6	Hex	11	Undec
2	Eth	7	Hept	12	dodec
3	Prop	8	Oct	13	tridec
4	But	9	Non		
5	Pent	10	Dec		

(d) **Primary suffix** :- According to saturation and unsaturation.



(e) **Secondary Suffix** :- According to senior most of F. G.



3-Formyl-4-hydroxy-2-methyl pentanoic acid

# CLASSIFICATION & NOMENCLATURE EXERCISE

S. NO.	Functional group	Prefix	Suffix
1.	—(C)OOH (carboxylic acid) —COOH	× carboxy	oic acid carboxylic acid
2.	—SO <sub>3</sub> H (sulphonic acid)	sulpho	sulphonic acid
3.	$\begin{array}{c} \text{O} \\ \parallel \\ \text{—(C)} \\ \text{—(C)} \end{array} > \text{O} \text{ (anhydride)}$	×	oic anhydride
4.	—(C)OOR (ester) —COOR	× alkoxycarbonyl or carbalkoxy	alkyl ----- oate alkyl ----- carboxylate
5.	—(C)OX (acid halide) —COX	× haloformyl	oyl halide carbonyl halide
6.	—(C)ONH <sub>2</sub> (amide) —CONH <sub>2</sub>	× carbamoyl	amide carboxamide
7.	—(C)N (cyanide) —CN	× cyano	Nitrile carbonitrile
8.	—N ≡ C (isocyanide)	isocyano/carbylamino	isonitrile/carbylamine
9.	—(C)HO (aldehyde) —CHO	oxo formyl	al carbaldehyde
10.	—(C)— (Ketone) $\begin{array}{c} \text{O} \\ \parallel \\ \text{—(C)—} \end{array}$	keto/oxo	one
11.	—OH (alcohol)	hydroxy	ol
12.	—SH (thio alcohol)	mercapto	thiol
13.	—NH <sub>2</sub> (amine)	amino	amine

**Note :** (C) atom written in brackets means that it has been included in the parent chain.

SUBSTITUENTS	PREFIX
—R	alkyl
—NH <sub>2</sub>	amino
—O—N=O	nitrito
—OCH <sub>2</sub> CH <sub>3</sub>	ethoxy
—CH <sub>2</sub> —Cl	chloromethyl
—S—	thio
$\text{CH}_3\text{—}\begin{array}{c} \text{C} \\ \parallel \\ \text{O} \end{array}\text{—O—}$	acetoxo/ethanoyloxy
$\text{C}_6\text{H}_5\text{—}\begin{array}{c} \text{C} \\ \parallel \\ \text{O} \end{array}\text{—O—}$	benzoyloxy

SUBSTITUENTS	PREFIX
—X	halo
$\text{—N} \begin{array}{l} \nearrow \text{O} \\ \searrow \text{O} \end{array}$	nitro
—N=O	nitroso
—CH <sub>2</sub> —OH	hydroxymethyl
—NH—CH <sub>3</sub>	methylamino
$\text{CH}_3\text{CH}_2\text{—}\begin{array}{c} \text{C} \\ \parallel \\ \text{O} \end{array}\text{—O—}$	propanoyloxy
—OR	Alkoxy
—OC <sub>6</sub> H <sub>5</sub>	Phenoxy